

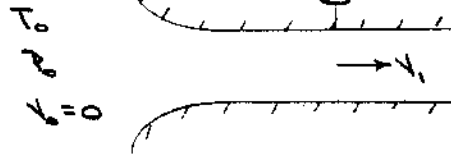
Problem 6.40

Given: Air flow in open circuit wind tunnel as shown.

$$P_{atm} - P_1 = 45 \text{ mm H}_2\text{O}$$

$$T_0 = 25^\circ\text{C}$$

$$P_0 = P_{atm}$$



Consider air to be incompressible.

Find: Air speed in tunnel at section 1

Solution:

Basic equations: $\frac{P}{\rho} + \frac{V^2}{2} + gz = \text{constant}$

- Assumptions:
- (1) steady flow
 - (2) incompressible flow
 - (3) frictionless flow
 - (4) flow along a streamline
 - (5) air behaves as an ideal gas
 - (6) stagnation pressure = P_{atm}

From the Bernoulli equation, $\frac{P_0}{\rho} = \frac{P_1}{\rho} + \frac{V_1^2}{2}$

$$P_0 - P_1 = P_{atm} - P_1 = \frac{1}{2} \rho V_1^2$$

$$V_1 = \left[\frac{2(P_{atm} - P_1)}{\rho} \right]^{1/2}$$

From the manometer reading, $P_{atm} - P_1 = \rho_{H_2O} g h$

$$V_1 = \left[\frac{2 \rho_{H_2O} g h}{\rho} \right]^{1/2}$$

From the ideal gas equation of state

$$\rho = \frac{P}{RT} = \frac{100 \times 10^3 \text{ N/m}^2 \times \frac{1 \text{ kg} \cdot \text{K}}{287 \text{ N} \cdot \text{m}}}{298 \text{ K}} = 1.17 \text{ kg/m}^3$$

$$V_1 = \left[\frac{2 \rho_{H_2O} g h}{\rho} \right]^{1/2} = \left[\frac{2 \times 999 \times 9.81 \text{ m/s}^2 \times 0.045 \text{ m}}{1.17} \right]^{1/2} = 27.5 \text{ m/s} \rightarrow V_1$$

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